

## Chapter 9 Mixed Review Stoichiometry

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Stoichiometry b. Theoretically, how many moles of  $\text{NH}_3$  will be produced? PROBLEMS Write the answer on the line to the left, Show all your work in the space provided. 1 88% The actual yield of a reaction is 22 g and the theoretical yield is 25 g. Calculate the percentage yield. 2. 6.0 mol of  $\text{N}_2$  are mixed with 12.0 mol of  $\text{H}_2$  according to the ...

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Reaction stoichiometry, the subject of this chapter, is based on chemical equations and the law of conservation of mass. All reaction stoichiometry calculations start with a balanced chemical equation. This equation gives the ... 290 Chapter 9 DO NOT EDIT--Changes must be made through "File info" ...

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